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PHOTO ACTIVITY OF MULTILAYER FILMS IN THE PRODUCTION OF METHANOL VIA CO2 REDUCTION

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This present work assessed the oxidative and reductive photocatalytic properties of powders fixed in the form of thick and porous TiO2 films in multilayers modified with niobium, tantalum and tungsten obtained by the Pechini method and the surface modification by the metallic Pt deposition through sputtering. The structural modification of TiO2 with these metals aimed at replacing Ti4+ by Nb5+, Ta5+ and W6+, which promoted, besides the increase in the electronhole pair recombination time, the generation of electron donor levels in the crystalline structure and the surface modification of the films with metallic Pt deposited by sputtering that acted in the conduction of photogenerated electrons. The films were characterized using the techniques PL and SEM/EDS to verify their morphology and the presence of characteristics defects. The photochemical assays of methanol formation from CO2 were performed in aqueous medium. To assess the methanol produce, samples were analyzed by GC to quantify the its formation. The best result achieved by the surface modification with Pt dots was observed on (Ti / SnO2 / Pt / TiO2500°C / Pt?) film that produced 6,2 ppm of methanol in 24 h reaction, concentration higher than achieved by the photocatalyst powder TiO2500°C (which has higher specific surface area) that reached 2,7 ppm.